|  |  |  |  |  |  |  |  |  |
| --- | --- | --- | --- | --- | --- | --- | --- | --- |
|  |  | **Non-Metals** | | | | | | |
| **F** 1- | **Br** | **O** 2- | **S** | **N** | **P** | **I** |
| **Metals** | **Li**  1+ | Lithium Fluoride  LiF |  | Lithium  Oxide  Li2O |  |  |  |  |
| **Na** |  |  |  |  |  |  |  |
| **K** |  |  |  |  |  |  |  |
| **Be** |  |  |  |  |  |  |  |
| **Mg** |  |  |  |  |  |  |  |
| **Ca** |  |  |  |  |  |  |  |
| **Al** |  |  |  |  |  |  |  |

**Directions:**1. For each element, write the oxidation number next to the symbol  
2. Then, fill in each box with the **name of the compound** and the **chemical formula** of the compound. There are a few done for you to serve as examples.